TWO APPROACHES TO THE SYNTHESIS OF 9-OXO- AND 10-HYDROXY-2*E***-DECENOIC ACIDS, IMPORTANT COMPONENTS OF QUEEN SUBSTANCE AND ROYAL JELLY OF HONEYBEES** *Apis mellifera*

G. Yu. Ishmuratov,¹ M. P. Yakovleva,¹ K. A. Tambovtsev,³ UDC 547.315+547.323+ Yu. V. Legostaeva,¹ **L. V. Kravchenko,**¹ **N. M. Ishmuratova,**¹ 547.391.8 **and G. A. Tolstikov2**

*Two approaches to the synthesis of 9-oxo- and 10-hydroxy-2*E*-decenoic acids, biologically active components of queen substance and royal jelly of honeybees, respectively, were proposed starting with allyl bromide and 1,7-octadiene and using chemo- and regioselective transformations of the common intermediate building block 7-octen-1-ylacetate.*

Key words: allyl bromide, 1,7-octadiene, 7-octen-1-ylacetate, 7-oxooct-1-ylacetate, 8-hydroxyoct-1-ylacetate, 9-oxo-2*E*decenoic acid, 10-hydroxy-2*E*-decenoic acid, synthesis.

9-Oxo- (**1**) and 10-hydroxy-2*E*-decenoic (**2**) acids have been identified as the most important components of queen substance and royal jelly, respectively, of the honeybee *Apis mellifera* L.

Oxoacid **1** is a polyfunctional pheromone of honeybees, plays the exclusive role of regulating their behavior and activity [1], and possesses significant pharmacological properties (antibacterial, anti-inflammatory, accelerator of graft wound and thermal burn healing, immunomodulator) [2] and antidote activity [3]. Compound **2** has bactericidal, fungicidal, and antitumor properties [4].

1) Institute of Organic Chemistry, Ufa Scientific Center, Russian Academy of Sciences, RF, 450054, Ufa, prosp. Oktyabrya, 71, fax (3472) 35 60 66, e-mail: insect@anrb.ru; 2) N. N. Vorozhtsov Novosibirsk Institute of Oragnic Chemistry, Siberian Division, Russian Academy of Sciences, 630090, Novosibirsk, posp. Akad. Lavrent′eva, 9; 3) Birsk State Social-Pedagogical Academy, RF, Republic of Bashkortostan, 452320, Birsk, ul. Internatsional′naya, 10. Translated from Khimiya Prirodnykh Soedinenii, No. 1, pp. 58-60, January-February, 2008. Original article submitted October 24, 2007.

Several syntheses of **1** and **2** have been reported. They differ in the methods of introducing oxo-, hydroxy-, and α , β -unsaturated carboxylic acid groups [5].

We propose two approaches to preparing acids **1** and **2** that are based on previously described transformations [6, 7] of keto-**7**- and aldehydo-**9**-acetates using the Doebner reaction for constructing the conjugated carboxylic acid. Intermediates **7** and **9**, in turn, were synthesized from available allyl bromide and 1,7-octadiene (**3**) [8] using their common key synthon 7-octen-1-ylacetate (**6**). For this, the former of these was first converted by a catalyzed (CuI:2,2′-bipyridyl) cross-conjugation reaction into the tetrahydropyran ether of 7-octen-1-ol (**4**) and then into the required acetate (**6**). The other approach was based on selective monohydroalumination of 3 by diisobutylaluminum hydride at room temperature in the presence of $ZrCl₄$ catalyst (in contrast with the previously reported [8] thermal version using triisobutylaluminum at 100°C) and oxidation of the resulting organaluminum compound to 7-octen-1-ol (**5**), which was then converted to acetate **6** as usual.

Further transformations of **6** in the direction of **1** consisted of its one-step Walker—Tsuji transformation into ketoacetate **7**. The building block of **9** was constructed for hydroxyacid **2** through a two-step synthesis using the intermediate monoester of 1,8-octanediol **8** based on chemo- and regioselective hydroboration—oxidation reactions.

EXPERIMENTAL

IR spectra were recorded on a UR-20 instrument as thin layers. PMR spectra in CDCl₃ were recorded on a Tesla BS-567 spectrometer (operating frequency 100 MHz) with TMS internal standard. GC was performed on Chrom-5 [column length 1.2 m, stationary phase silicone SE-30 (5%) on Chromaton N-AW-DMCS (0.16-0.20 mm), operating temperature 50- 300°C] and Chrom-41 [column length 2.4 m, stationary phase PEG-6000, operating temperature 50-200°C] instruments with He carrier gas. Column chromatography used SiO_2 (70-230, Lancaster, England). TLC monitoring used SiO_2 (Sorbfil, Russia). Elemental analyses of all compounds agreed with those calculated.

1-(2-Tetrahydropyranyloxy)-7-octene (4). A suspension of CuI (2.85 g, 15.0 mmol) in absolute THF (58 mL) was treated with 2,2′-bipyridine (2.34 g, 15.0 mmol), stirred for 0.5 h (20°C, Ar), cooled to 2°C, treated with allylbromide (10.41 g, 86.0 mmol) in absolute THF (17 mL), stirred for 10 min, treated with Grignard reagent prepared from Mg (1.11 g, 46.0 mg-at) and 1-(2-tetrahydropyranyloxy)-5-bromopentane (**2**) as before [9] in THF (40.5 mL), stirred for 1 h at 10°C, and treated with Et₂O (500 mL). The organic layer was separated, washed with saturated NaCl solution, dried over Na₂SO₄, and evaporated. The residue was chromatographed (SiO₂, hexane:Et₂O, 15:1) to afford **4** (5.82 g, 69%). IR spectrum (v, cm⁻¹): 3080, 1650, 920 (CH=CH2), 1150, 1090, 1045 (C–O–C).

7-Octen-1-ol (5). A solution of 1,7-octadiene (**3**, 5.00 g, 45.5 mmol) prepared as before [8] in absolute hexane (45 mL) was stirred (20 $^{\circ}$ C, Ar), treated with $ZrCl₄$ (0.29 g, 1.25 mmol) and diisobutylaluminum hydride (73% solution, 11.5 mL, 45.9 mmol) in toluene, stirred for 7 h at 20°C, purged successively with dry air (0-20°C, 1 h) and oxygen (20°C, 1 h, 30-40°C, 3 h), cooled to 0 \degree C, treated dropwise with H₂SO₄ (38 mL, 10%), stirred for 1 h at room temperature, and extracted with Et₂O (3 × 100 mL). The combined extracts were washed successively with saturated solutions of NaHCO₃ and NaCl, dried over Na₂SO₄, evaporated, and distilled to afford **5** (4.30 g, 74%), bp 64-66°C (7 mm). IR spectrum (v, cm^{−1}): 3550 (OH), 3090, 1640, 920 (CH=CH2). PMR spectrum (δ, ppm): 1.20-1.70 (8H, m, H-2—H-5), 2.15-2.30 (2H, m, H-6), 3.66 (2H, t, J = 7 Hz, H-1), 5.0-5.4 (2H, m, H-8), 5.6-5.9 (1H, m, H-7), similar to the literature data [10].

7-Octen-1-ylacetate (6), a: ether **4** (5.75 g, 27.1 mmol) was treated with a mixture of AcOH and AcCl (10:1, 27 mL), stored for 48 h at 30-40°C, diluted with Et₂O (250 mL), washed successively with saturated solutions of NaHCO₃ and NaCl, dried over MgSO₄, and evaporated to afford **6** (3.41 g, 74%). IR spectrum (v, cm^{−1}): 3090, 1645, 920 (CH=CH₂), 1745, 1250 (OAc). PMR spectrum (δ, ppm): 1.30-1.75 (8H, m, H-2—H-5), 2.05 (3H, s, CH₃CO), 2.15-2.30 (2H, m, H-6), 4.06 (2H, t, $J = 6.5$ Hz, H-1), $5.00 - 5.40$ (2H, m, H-8), $5.60 - 5.90$ (1H, m, H-7).

b: a mixture of 5 (4.00 g, 31.3 mmol), dry Py (63.7 mL), and Ac₂O (27.3 mL) was stored for 24 h at room temperature and evaporated. The residue was dissolved in Et₂O (250 mL), washed with HCl (10%) and saturated solutions of NaHCO₃ and NaCl, dried over MgSO₄, and evaporated. The residue was chromatographed (SiO₂, hexane:Et₂O, 9:1) to afford **6** (4.30 g, 81%), spectral properties identical to those for the compound in **a**.

7-Oxooct-1-ylacetate (7). A mixture of PdCl₂ (0.38 g, 2.1 mmol), Cu₂Cl₂ (2.19 g, 11.0 mmol), DMF (11.4 mL), and $H₂O$ (1.4 mL) was stirred for 1 h under $O₂$, treated with 6 (3.40 g, 20.0 mmol), stirred for 6 h to absorb $O₂$ (270 mL), diluted with CHCl₃ (250 mL), washed successively with HCl (5%) and saturated NaCl solution, dried over Na₂SO₄, and evaporated to afford **7** (2.42 g, 65%). The IR and PMR spectra were identical to those reported earlier [11, 12].

8-Hydroxyoct-1-ylacetate (8). A suspension of 9-BBN (3.55 g, 29.1 mmol) in absolute THF (36 mL) at 10°C was treated with a solution of **7** (3.50 g, 20.6 mmol) in absolute THF (12 mL), stored for 2 h, cooled to 0°C, treated with a solution of NaOAc (5.92 g, 72.2 mmol) in H₂O (14 mL) and dropwise with H₂O₂ (30%, 21.3 mL), stirred for 2 h at room temperature, diluted with Et₂O (300 mL), washed successively with saturated NaCl solution, Na₂S₂O₃ solution (0.1 N), again with saturated NaCl solution, dried over Na₂SO₄, and evaporated. The residue was chromatographed (SiO₂, hexane:ether, 10:1) to afford 8 (2.75 g, 71%). The IR and PMR spectra were identical to those reported earlier [13].

8-Acetoxyoctanal (9). A suspension of PCC (5.32 g, 24.7 mmol) in dry CH₂Cl₂ (28 mL, Ar, 20°C) was treated in one portion with a solution of **8** (2.70 g, 14.4 mmol) in dry CH₂Cl₂ (3 mL), diluted after 1.5 h with Et₂O (35 mL), and filtered through a layer of $SiO₂$ (15 cm). The solid was washed with Et₂O. The combined filtrate was evaporated to afford **9** (2.38 g, 89%). The IR and PMR spectra were identical to those reported earlier [7].

10-Hydroxy-2*E***-decenoic acid (2)** was prepared from **9** in two steps in overall yield 49% as before [14], mp 64-65°C [15]. The IR and PMR spectra were identical to those reported earlier [14].

9-Oxo-2*E***-decenoic acid (1)** was prepared from 7-oxooct-1-ylacetate (**7**) in three steps in overall yield 34% as before [15], mp 52-54°C [16]. The IR and PMR spectra were identical to those reported earlier [17].

REFERENCES

- 1. K. V. Lebedeva, V. A. Minyailo, and Yu. B. Pyatnova, *Insect Pheromones* [in Russian], Nauka, Moscow (1984).
- 2. G. Yu. Ishmuratov, A. F. Ismagilova, A. A. Sharipov, O. N. Gerasyuta, R. Ya. Kharisov, N. M. Ishmuratova, and G. A. Tolstikov, *Khim.-farm. Zh.*, **37**, 31 (2003).
- 3. N. M. Ishmuratova, G. Yu. Ishmuratov, G. A. Tolstikov, A. F. Ismagilova, and A. E. Belov, *Vestn. Ross. Akad. Skh. Nauk*, 84 (2007).
- 4. Yu. B. Pyatnova, L. L. Ivanov, and A. S. Kyskina, *Usp. Khim.*, **38**, 248 (1969).
- 5. G. Yu. Ishmuratov, R. Ya. Kharisov, O. V. Botsman, N. M. Ishmuratova, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 3 (2002).
- 6. G. Yu. Ishmuratov, M. P. Yakovleva, L. P. Botsman, N. M. Ishmuratova, R. R. Muslukhov, G. V. Khambalova, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 28 (2003).
- 7. R. Chiron, *J. Chem. Ecol.*, **8**, 709 (1982).
- 8. G. A. Tolstikov, V. N. Odinokov, R. I. Galeeva, R. S. Bakeeva, and V. R. Akhunova, *Khim. Prir. Soedin.*, 239 (1982).
- 9. V. N. Odinokov, G. Yu. Ishmuratov, L. P. Botsman, R. R. Vakhidov, I. M. Ladenkova, T. A. Kargapol'tseva, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 423 (1992).
- 10. H. Matsuda, A. Yamamoto, N. Iwamoto, and S. Matsuda, *J. Org. Chem.*, **43**, 4567 (1978).
- 11. G. Yu. Ishmuratov, M. P. Yakovleva, L. P. Botsman, N. M. Ishmuratova, R. R. Muslukhov, G. V. Khambalova, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 28 (2003).
- 12. G. Yu. Ishmuratov, M. P. Yakovleva, G. V. Zaripova, A. V. Galyautdinova, L. P. Botsman, and N. M. Ishmuratova, *Bashk. Khim. Zh.*, 36 (2004).
- 13. Y. Katsumi, *J. Chem. Soc. Jpn. Chem. Ind.*, 1415 (1978).
- 14. V. N. Odinokov, G. Yu. Ishmuratov, and G. A. Tolstikov, *Khim. Prir. Soedin.*, 695 (1983).
- 15. Yu. K. Pyatraitis, *Khemoretseptsiya Nasekomykh*, 31 (1978).
- 16. Yu. K. Pyatraitis, *Khemoretseptsiya Nasekomykh*, 213 (1975).
- 17. Yu. K. Pyatraitis and V. K. Daukshas, *Khemoretseptsiya Nasekomykh*, 55 (1980).